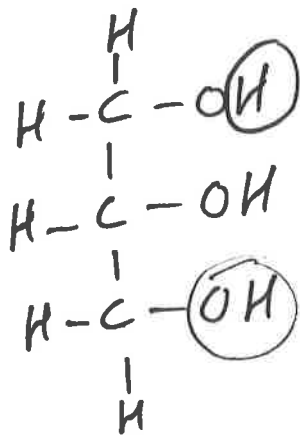
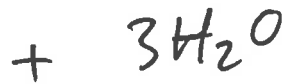
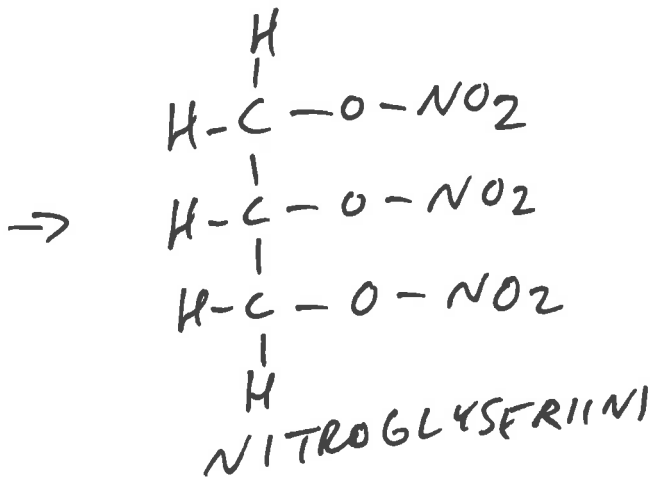
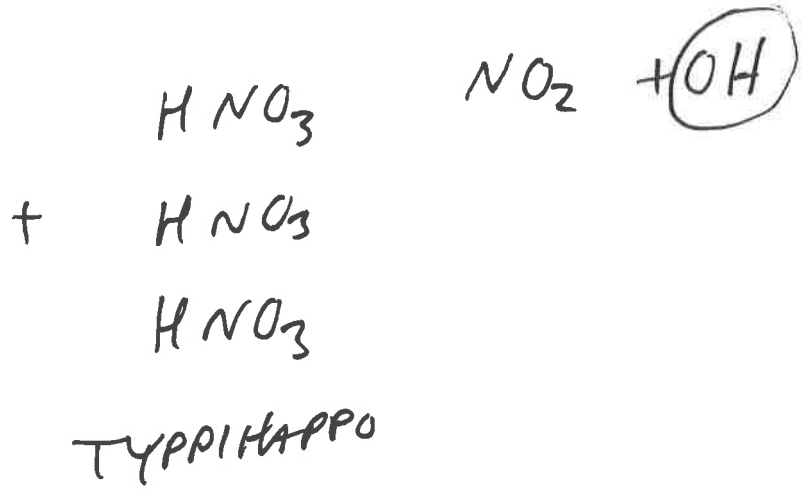


$$V_m = 22,4 \text{ L}$$

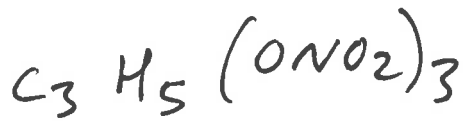
ESIM. NITROGLYSERIINI RÄJÄHTÄÄ



GLYSEROLI



VETÄ

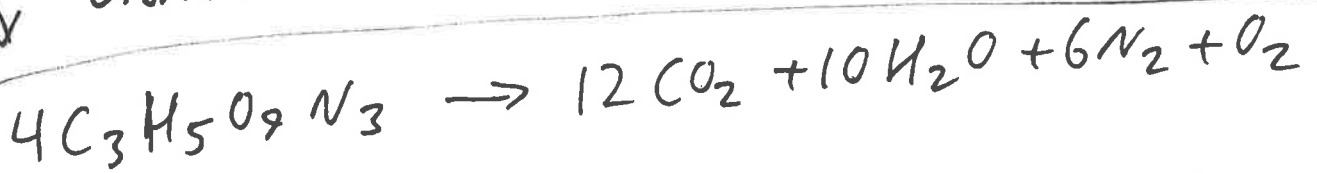




C_3
m

C_1
m

TASAPAINOTETAAN



HUOM! RÄJÄHDYS EI VAADI HAPPEA!
SE TUOTTA HAPPEA.

KYSYMYS: JOS 4 MOOLIA $\text{C}_3\text{H}_5\text{O}_9\text{N}_3$
RÄJÄHTÄÄ, MONTAKO LITRAA
TULEE LOPPUTUOTTEITA?

RATK. LOPPUTUOTTEITA $12 + 10 + 6 + 1$
 $= 29$ MOOLIA

$$1 \text{ MOOLI KAASUA} = V_m = 22,4 \text{ L}$$

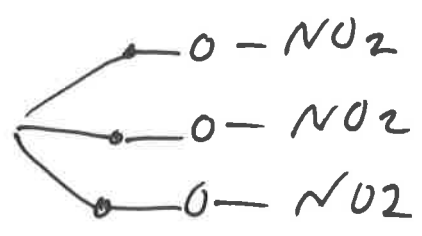
$$V = 29 \cdot 22,4 \text{ L} \\ = \underline{\underline{638 \text{ L}}}$$

• PALJONKO NITROGLYSERIINIÄ
TARVIKSIIN, $m = ?$

$$M(\text{C}_3\text{H}_5\text{O}_9\text{N}_3) = 3 \cdot 12 + 5 \cdot 1 + 9 \cdot 16 + 3 \cdot 14 \\ = 36 + 5 + 144 + 42 \\ = 227 \frac{\text{g}}{\text{MOL}}$$

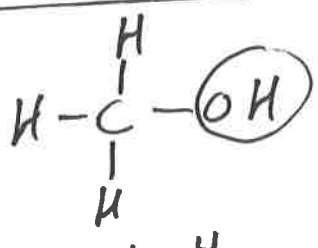
$$m = n M = 4 \text{ mol} \cdot 227 \frac{\text{g}}{\text{mol}} \\ = 2 \cdot 452 = \underline{\underline{900 \text{ g}}}$$

ENERGIAA VAPAUTUU, KOSKA

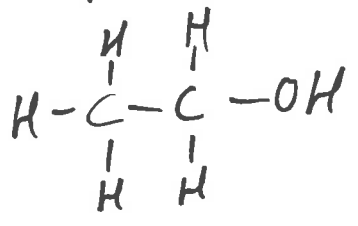


↑
TODELLA ALHAISEN
ENERGIAN SIDOS

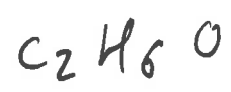
ALKOHOLI



METANOLI



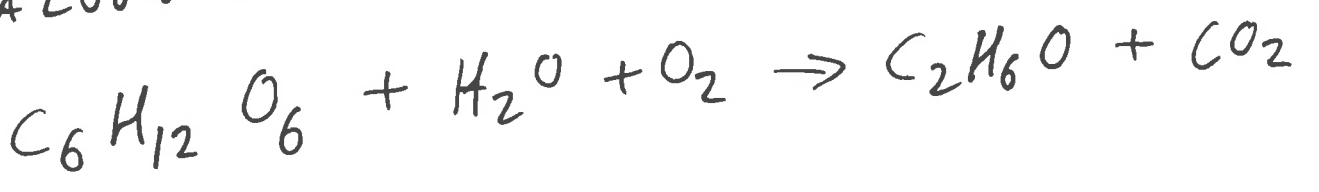
ETANOLI



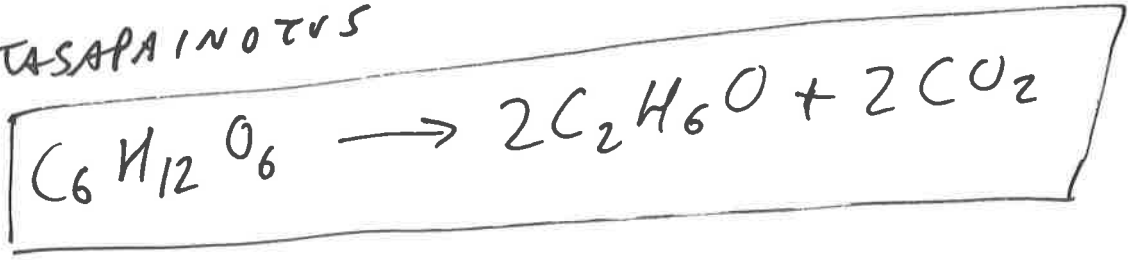
ESIM.

10 L VETTA' → 10 L JUOMAA
 PAKETTI HIIVAA → 12% ETANOLIA
 X kg GLUKOOSIA → 88% VETTA'

PALJONKO GLUKOOSIA TARVIKSIAN ?



TASAPAINOTUS



$$E\text{TANOLIA } 12\% \rightarrow 0,12 \cdot 10\text{ L} = 1,2\text{ L}$$

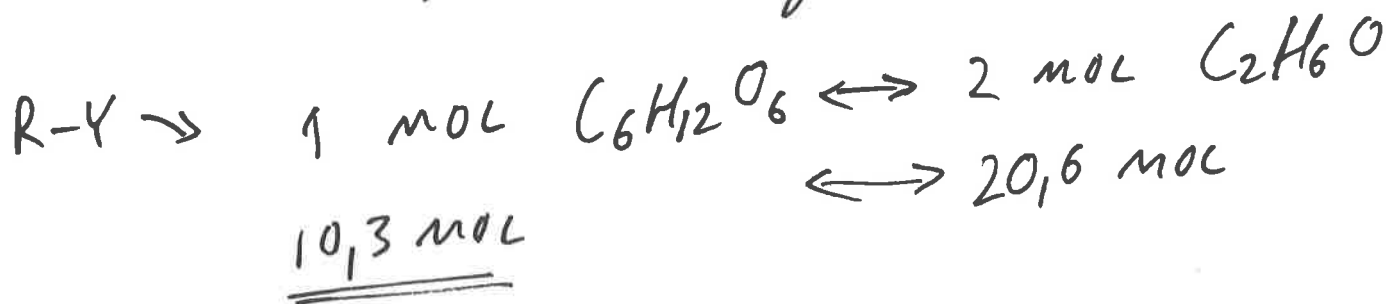
$$V = 10\text{ L}$$

$$\rho(\text{ETANOLI}) = 0,79\text{ kg/L}$$

$$m(\text{ETANOLI}) = 1,2\text{ L} \cdot 0,79 \frac{\text{kg}}{\text{L}} = 0,948\text{ kg}$$
$$= 948\text{ g}$$

$$M(\text{C}_2\text{H}_6\text{O}) = 2 \cdot 12 + 6 \cdot 1 + 16$$
$$= 24 + 6 + 16 = 46 \frac{\text{g}}{\text{mol}}$$

$$n(\text{ETANOLI}) = \frac{m}{M} = \frac{948\text{ g}}{46\text{ g/mol}} = 20,6\text{ mol}$$



$$M(\text{C}_6\text{H}_{12}\text{O}_6) = 6 \cdot 12 + 12 \cdot 1 + 6 \cdot 16$$
$$= 180 \frac{\text{g}}{\text{mol}}$$

$$m(\text{C}_6\text{H}_{12}\text{O}_6) = n M = 10,3 \cdot 180\text{ g}$$
$$= \underline{\underline{1,85\text{ kg}}}$$

RUOKOSOKERI = 50% GLUKOOSIA + 50% FRUKTOOSIA

$$\text{JÄRVITÄÄN } 2 \cdot 1,85\text{ kg} = \underline{\underline{3,7\text{ kg}}}$$

HILUIDIOKSIDIA SYNTYÄ

$$V = 20,6\text{ mol} \cdot V_m$$

$$= 20,6 \cdot 22,4\text{ L} = \underline{\underline{450\text{ L}}}$$